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The reaction

$AB(aq) \rightarrow A(g) + B(g)$  is second order in AB and has a rate constant of  $0.0114 \text{ L} \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$  at  $25.0 \text{ }^\circ\text{C}$ . A reaction vessel initially contains  $250.0 \text{ mL}$  of  $0.135 \text{ mol} \cdot \text{L}^{-1}$  AB which is allowed to react to form the gaseous product. The product is collected over water at  $25.0 \text{ }^\circ\text{C}$ .

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which solid and liquid  
phases exist in dynamic  
equilibrium is called  
\_\_\_\_\_. a. the sublimation  
point of the solid. b. the  
melting point of the solid  
or freezing point of the  
liquid. c. the boiling point  
of the liquid. d. the vapor  
pressure of the liquid.

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The reaction  
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is allowed to react to form  
the gaseous product. The  
product is collected over

water at  $25.0 \text{ }^\circ\text{C}$ .

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