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Ans: Class 12 Chemistry Chapter 7 discusses the properties of the p-Block elements. The p-Block elements are the elements from Group 13 to 18 in the periodic table. The study of the elements and their electronic configuration is an important part of Class12 Chemistry Part 1.

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7.1 Introduction: Recall from Chapter 1 that solutions are defined as homogeneous mixtures that are mixed so thoroughly that neither component can be observed independently of the other. Solutions are all around us. Air, for example, is a solution. If you live near a lake, a river, or an ocean, that body of water is not pure H_2O but most probably a solution.

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The Phase Diagram of Water. Figure 7.7.2 shows the phase diagram of water and illustrates that the triple point of water occurs at 0.01°C and 0.00604 atm (4.59 mmHg). Far more reproducible than the melting point of ice, which depends on the amount of dissolved air and the atmospheric pressure, the triple point (273.16 K) is used to define the absolute (Kelvin) temperature scale.

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