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5. Amongst the following chemical reactions, the irreversible reaction is: a)  $\text{AgNO}_3 + \text{NaCl} \rightleftharpoons \text{AgCl} + \text{NaNO}_3$ . b)  $\text{H}_2 + \text{I}_2 \rightleftharpoons \text{HI}$ . c)  $\text{CaCO}_3 \rightleftharpoons \text{CaO} + \text{CO}_2$ . d)  $\text{O}_2 + 2\text{SO}_2 \rightleftharpoons 2\text{SO}_3$ . ANS: The reaction is not reversible. 6. When the rate of forward reaction becomes equal to backward reaction, this state is termed as: a) Chemical equilibrium. b) Reversible state

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chemical equation for the reaction at equilibrium is written using double arrows to indicate the overall reversibility of the reaction.  $2\text{HgO}(s) \rightleftharpoons 2\text{Hg}(l) + \text{O}_2(g)$  Equilibrium, a Dynamic State Many chemical reactions are reversible under ordinary conditions of temperature and concentration. They will reach a state of equilibrium

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562 Chapter 18 Chemical Equilibrium Figure 18-3 If the only way in or out of San Francisco and Sausalito is across the Golden Gate Bridge, which joins the two cities, then the number of vehicles in the two cities will remain constant if the number of vehicles per hour crossing the bridge in one direction equals the number of vehi-

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